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MATERIAL SAFETY DATA SHEET

According to the UN GHS revision 9

Version: 1.0

Creation Date: July 15, 2019 Revision Date: July 15, 2019

SECTION 1: Identification

1.1GHS Product identifier

Product name Aluminium sulphate

1.20ther means of identification

Product number

Other names Aluminium sulfate; Aluminum Sulfate;

1.3Recommended use of the chemical and restrictions on use

Industrial and scientific research use. **Identified uses**

Uses advised against no data available

1.4Supplier's details

Shandong Yili-Spring Chemical Industry Co., Ltd. Company

1016, Xinyue Fortune Center, Zouping County, Binzhou City, Shandong Address

86-543-4865599 Telephone

1.5Emergency phone number

Emergency phone number +86 15505433527

Service hours Monday to Friday, 8am-5pm (Standard time zone: UTC/GMT +8 hours).

SECTION 2: Hazard identification

2.1 Classification of the substance or mixture

Corrosive to metals, Category 1 Serious eye damage, Category 1

2.2GHS label elements, including precautionary statements



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Pictogram(s)



Signal word Danger

Hazard statement(s) H290 May be corrosive to metals

H318 Causes serious eye damage

Precautionary statement(s)

Prevention P234 Keep only in original packaging.

P280 Wear protective gloves/protective clothing/eye protection/face

protection/hearing protection/...

Response P390 Absorb spillage to prevent material damage.

P305+P354+P338 IF IN EYES: Immediately rinse with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P317 Get medical help.

Storage P406 Store in a corrosion resistant/...container with a resistant inner liner.

Disposal none

2.30ther hazards which do not result in classification

no data available

SECTION 3: Composition/information on ingredients

3.1Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
Aluminium sulphate	Aluminium sulphate	10043-01-3	233-135-0	100%

SECTION 4: First-aid measures

4.1Description of necessary first-aid measures

If inhaled

Fresh air, rest. Refer for medical attention.

Following skin contact

Rinse skin with plenty of water or shower.

Following eye contact

Rinse with plenty of water for several minutes (remove contact lenses if easily possible). Refer immediately for medical attention.



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Following ingestion

Rinse mouth. Do NOT induce vomiting. Give one or two glasses of water to drink. Refer for medical attention .

4.2Most important symptoms/effects, acute and delayed

Inhalation of dust irritates nose and mouth. Ingestion of large doses causes gastric irritation, nausea, vomiting, and purging. Dust irritates eyes and skin. (USCG, 1999)

Vapor irritates eyes, nose and respiratory tract due to formation of sulfuric acid. Ingestion of large doses causes gastric irritation, nausea, vomiting, and purging. Liquid irritates eyes and skin. (USCG, 1999)

4.3Indication of immediate medical attention and special treatment needed, if necessary

Exposure to aluminum sulfate/: Call for medical aid. Dust: Irritating to eyes, nose and throat. If inhaled will cause difficult breathing. If in eyes, hold eyelids open and flush with plenty of water. If breathing has stopped, give artificial respiration. If breathing is difficult, give oxygen. Solid: Irritating to skin and eyes. If swallowed with cause nausea or vomiting. Remove contaminated clothing and shoes. Flush affected areas with plenty of water. If in eyes, hold eyelids open and flush with plenty of water. If swallowed and victim is conscious, have victim drink water or milk. If swallowed and victim is unconscius or having convulsions, do nothing except keep victim warm.

SECTION 5: Fire-fighting measures

5.1Suitable extinguishing media

If material involved in fire: Extinguish fire using agent suitable for type of surrounding fire. (Material itself does not burn or burns with difficulty.)

5.2Specific hazards arising from the chemical

Special Hazards of Combustion Products: Produces sulfuric acid upon decomposition. (USCG, 1999)

5.3 Special protective actions for fire-fighters

In case of fire in the surroundings, use appropriate extinguishing media.

SECTION 6: Accidental release measures

6.1Personal precautions, protective equipment and emergency procedures

Personal protection: particulate filter respirator adapted to the airborne concentration of the substance. Do NOT let this chemical enter the environment. Sweep spilled substance into covered plastic containers. If appropriate, moisten first to prevent dusting. Store and dispose of according to local regulations.

6.2Environmental precautions

Personal protection: particulate filter respirator adapted to the airborne concentration of the substance. Do NOT let this chemical enter the environment. Sweep spilled substance into covered plastic containers. If appropriate, moisten first to prevent dusting. Store and dispose of according to local regulations.



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6.3Methods and materials for containment and cleaning up

Environmental considerations: Land spill: Dig a pit, pond, lagoon, holding area to contain liquid or solid material. /SRP: If time permits, pits, ponds, lagoons, soak holes, or holding areas should be sealed with an impermeable flexible membrane liner./ Cover solids with a plastic sheet to prevent dissolving in rain or fire fighting water.

SECTION 7: Handling and storage

7.1Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

7.2 Conditions for safe storage, including any incompatibilities

Separated from bases and strong oxidants. Dry. Store in an area without drain or sewer access. Provision to contain effluent from fire extinguishing. Ambient Octadecohydrate

SECTION 8: Exposure controls/personal protection

8.1Control parameters

Occupational Exposure limit values

Component	Aluminium	sulphate			
CAS No.	10043-01-3	10043-01-3			
	Limit value	Limit value - Eight hours Limit value - Short term			
	ppm	mg/m³	ppm	mg/m³	
Finland		1 (1)			
	Remarks				
Finland	(1) calculate	ed as Al			

Biological limit values

no data available

8.2Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

8.3Individual protection measures, such as personal protective equipment (PPE)

Eye/face protection

Wear safety goggles.



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Skin protection

Protective gloves.

Respiratory protection

Avoid inhalation of dust. Use local exhaust or breathing protection.

Thermal hazards

no data available

SECTION 9: Physical and chemical properties and safety characteristics

Physical state Solid. Crystalline.

Colour White, lustrous crystals, pieces, granules, or powder.

Odour ODORLESS

Melting point/freezing point > 770 °C. Atm. press.:Ca. 101 325 Pa.

Boiling point or initial boiling 330°C at 760 mmHg

point and boiling range

Flammability Not combustible. Gives off irritating or toxic fumes (or gases) in a fire.

Lower and upper explosion

no data available

limit/flammability limit

Flash point no data available

Auto-ignition temperature no data available

Decomposition temperature 770°C

pH > 1 - <= 2.9. Remarks:1 g/1 ml water) not less than 2.9.

Kinematic viscosity dynamic viscosity (in mPa s) = 17. Temperature:20°C. Remarks:Water

solution of aluminium sulphate 1 g/1 ml water.

Solubility Freely soluble in water, insoluble in ethanol

Partition coefficient no data available

n-octanol/water

Vapour pressure Esentially zero.

Density and/or relative density 2 710 kg/m³. Temperature:20 °C.

Relative vapour density no data available **Particle characteristics** no data available

SECTION 10: Stability and reactivity

10.1Reactivity

Decomposes on heating and on burning. This produces toxic and corrosive fumes including sulfur oxides. Reacts with bases. Reacts violently with strong oxidants. This produces heat. This produces toxic and corrosive



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fumes including sulfur oxides. The solution in water is a medium strong acid. Attacks many metals in the presence of water.

10.2Chemical stability

Stable in air.

10.3Possibility of hazardous reactions

May burn, but will not ignite. Aqueous solutions of ALUMINUM SULFATE are acidic. The solid may corrode metals in presence of moisture.

10.4Conditions to avoid

no data available

10.5Incompatible materials

May corrode metals in presence of moisture. Aluminum sulfate octadecahydrate

10.6Hazardous decomposition products

When heated to decomposition it emits toxic fumes of /sulfur oxides/.

SECTION 11: Toxicological information

Acute toxicity

Oral: LD50 Rat oral 1930 mg/kg

• Inhalation: no data available

• Dermal: no data available

Skin corrosion/irritation

no data available

Serious eye damage/irritation

no data available

Respiratory or skin sensitization

no data available

Germ cell mutagenicity

no data available

Carcinogenicity

A4: Not classifiable as a human carcinogen. Aluminum metal and insoluble compounds

Reproductive toxicity



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no data available

STOT-single exposure

The substance is severely irritating to the eyes, respiratory tract and gastrointestinal tract. The substance is mildly irritating to the skin.

STOT-repeated exposure

The substance may have effects on the central nervous system. This may result in impaired functions.

Aspiration hazard

A harmful concentration of airborne particles can be reached quickly when dispersed, especially if powdered.

SECTION 12: Ecological information

12.1Toxicity

- Toxicity to fish: NOEC Danio rerio (previous name: Brachydanio rerio) > 1 000 mg/L 96 h. Remarks:Low toxicity.
- Toxicity to daphnia and other aquatic invertebrates: EC50 Daphnia magna > 100 mg/L 48 h. Remarks: And immobilization.
- Toxicity to algae: EC50 Pseudokirchneriella subcapitata (previous names: Raphidocelis subcapitata, Selenastrum capricornutum) -> 100 mg/L 72 h.
- Toxicity to microorganisms: EC50 Euglena sp. 3 010.832 mg/L 5 d. Remarks: Population.

12.2Persistence and degradability

no data available

12.3Bioaccumulative potential

no data available

12.4Mobility in soil

no data available

12.5Other adverse effects

no data available

SECTION 13: Disposal considerations

13.1Disposal methods

Product



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The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

SECTION 14: Transport information

14.1UN Number

ADR/RID: Not dangerous goods. (For IMDG: Not dangerous goods. (For reference only, please check.)

reference only, please check.)

reference only, please check.)

reference only, please check.)

14.2UN Proper Shipping Name

ADR/RID: Not dangerous goods. (For IMDG: Not dangerous goods. (For reference only, please check.)

reference only, please check.)

reference only, please check.)

14.3Transport hazard class(es)

ADR/RID: Not dangerous goods. (For IMDG: Not dangerous goods. (For reference only, please check.)

reference only, please check.)

reference only, please check.)

reference only, please check.)

14.4Packing group, if applicable

ADR/RID: Not dangerous goods. (For IMDG: Not dangerous goods. (For reference only, please check.)

reference only, please check.)

reference only, please check.)

reference only, please check.)

14.5Environmental hazards

ADR/RID: No IMDG: No IATA: No

14.6Special precautions for user

no data available

14.7Transport in bulk according to IMO instruments

no data available

SECTION 15: Regulatory information

15.1Safety, health and environmental regulations specific for the product in question

Chemical name Common names and synonyms	CAS number	EC number	
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Aluminium sulphate	Aluminium sulphate 10043-01-3		233-135-0
European Inventory of Existing Commercial Chemical Substances (EINECS)			Listed.
EC Inventory			Listed.
United States Toxic Substances Control Act (TSCA) Inventory			Listed.
China Catalog of Hazardous chemicals 2015			Not Listed.
New Zealand Inventory of Chemicals (NZIoC)			Listed.
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			Listed.
Vietnam National Chemical Inventory			Listed.
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)			Listed.
Korea Existing Chemicals List (KECL)			Listed.

SECTION 16: Other information

Information on revision

Creation Date July 15, 2019 **Revision Date** July 15, 2019

Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

References

- IPCS The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home
- HSDB Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm
- IARC International Agency for Research on Cancer, website: http://www.iarc.fr/
- eChemPortal The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request locale=en
- CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple
- ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp
- ERG Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg



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Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp

• ECHA - European Chemicals Agency, website: https://echa.europa.eu/

Other Information

Occurs in nature as the mineral alunogenite. Other CAS numbers: 16828-12-9 (14-hydrate); 16828-11-8 (16-hydrate); 7784-31-8 (18-hydrate); 17927-65-0 (x-hydrate). Aluminium sulfate hydrolizes in water forming sulfuric acid and heat. Literature values for solubility of the substance are very different due to hydrolyzation process.

Disclaimer: The above information is believed to be correct but does not purport to be all inclusive and shall be used only as a guide. The information in this document is based on the present state of our knowledge and is applicable to the product with regard to appropriate safety precautions. It does not represent any guarantee of the properties of the product. We as supplier shall not be held liable for any damage resulting from handling or from contact with the above product.